

#### **Product Data Sheet**

## AMBERLITE™ HPR4811 Cl Ion Exchange Resin

Uniform Particle Size, Gel, Strong Base Anion Exchange Resin for Industrial Demineralization Applications

### **Description**

AMBERLITE™ HPR4811 CI Ion Exchange Resin is a high-quality resin for use in industrial demineralization applications when high performance and cost-effective operation is required. The chemical properties and particle size of the resin have been optimized to help yield excellent operating capacity and rinse characteristics, while reducing chemical regenerant and rinse water usage.

AMBERLITE™ HPR4811 CI provides high capacity with excellent kinetics and very good resistance to organic fouling. It also has excellent resistance to osmotic shock and good physical and chemical stability. Its unique resistance to organic fouling enables this resin to be used in CI-form as an organic scavenger with great success in a single bed unit.

## **Applications**

- Demineralization
  - Ideally when treating water with:
    - High organic fouling potential
    - High percentage of silica
  - When the treatment goal is:
    - Removal of strong and weak acids
    - Lowest silica leakage
- Scavenging

### **System Designs**

- Co-current
- Counter-current / Hold-down
- Packed beds

# Historical Reference

AMBERLITE™ HPR4811 CI Ion Exchange Resin has previously been sold as DOWEX MARATHON™ 11 Ion Exchange Resin.

Page 1 of 4 Form No. 177-03792, Rev. 1

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# **Typical Properties**

Physical Properties	
Copolymer	Styrene-divinylbenzene
Matrix	Gel
Туре	Strong base anion, Type I
Functional Group	Trimethylammonium
Physical Form	Amber, translucent, spherical beads
Chemical Properties	
Ionic Form as Shipped	Cl
Total Exchange Capacity	≥ 1.3 eq/L (Cl <sup>-</sup> form)
Water Retention Capacity	50.0 – 60.0% (Cl <sup>-</sup> form)
Particle Size §	
Particle Diameter	$550 \pm 50  \mu m$
Uniformity Coefficient	≤1.10
< 300 µm	≤0.3%
> 850 µm	≤ 1.0%
Stability	
Whole Uncracked Beads	≥95%
Swelling	$Cl^- \rightarrow OH^-: 20\%$
Density	
Particle Density	1.08 g/mL
Shipping Weight	675 g/L

<sup>§</sup> For additional particle size information, please refer to the Particle Size Distribution Cross Reference Chart (Form No. 177-01775).

# Suggested Operating Conditions

Temperature Range		
OH <sup>-</sup> form <sup>‡</sup>	5-60°C (41-140°F)	
Cl <sup>⁻</sup> form	5-100°C (41-212°F)	
pH Range		
Service Cycle (demineralization)	1 – 14	
Service Cycle (scavenging)	2-10	
Stable	0 – 14	

<sup>&</sup>lt;sup>‡</sup> Operating at elevated temperatures, for example above 60 – 70°C (140 – 158°F), may impact resin life. Contact our technical representative for details.

For additional information regarding recommended minimum bed depth, operating conditions, and regeneration conditions for <u>separate beds</u> (Form No. 177-03729) in water treatment, please refer to our Tech Fact.

Page 2 of 4 Form No. 177-03792, Rev. 1

### Hydraulic Characteristics

Estimated bed expansion of AMBERLITE™ HPR4811 CI Ion Exchange Resin as a function of backwash flowrate and temperature is shown in Figure 1.

Estimated pressure drop for AMBERLITE™ HPR4811 CI as a function of service flowrate and temperature is shown in Figure 2. These pressure drop expectations are valid at the start of the service run with clean water.

Figure 1: Backwash Expansion

Temperature =  $10 - 60^{\circ}\text{C} (50 - 140^{\circ}\text{F})$ 

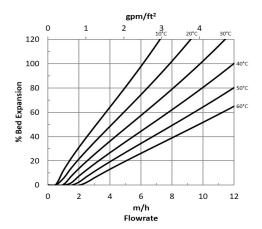
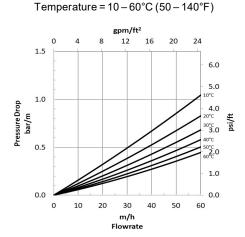


Figure 2: Pressure Drop



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### **Customer Notice**

DuPont strongly encourages its customers to review both their manufacturing processes and their applications of DuPont products from the standpoint of human health and environmental quality to ensure that DuPont products are not used in ways for which they are not intended or tested. DuPont personnel are available to answer your questions and to provide reasonable technical support. DuPont product literature, including safety data sheets, should be consulted prior to use of DuPont products. Current safety data sheets are available from DuPont.

Please be aware of the following:

• WARNING: Oxidizing agents such as nitric acid attack organic ion exchange resins under certain conditions. This could lead to anything from slight resin degradation to a violent exothermic reaction (explosion). Before using strong oxidizing agents, consult sources knowledgeable in handling such materials.

Page 3 of 4 Form No. 177-03792, Rev. 1

#### Have a question? Contact us at:

www.dupont.com/water/contact-us

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Page 4 of 4 Form No. 177-03792, Rev. 1