

Product Data Sheet

AMBERLYST™ 36DRY Polymeric Catalyst

Industrial-grade, Strongly Acidic Catalyst

Description

AMBERLYST™ 36DRY Polymeric Catalyst is a bead-form, macroporous, sulfonic acid resin developed particularly for heterogeneous catalysis.

The special process used to manufacture AMBERLYST™ 36DRY results in a particularly high concentration of acid groups and also confers an improved thermal stability when compared to catalysts with a similar level of crosslinking. It is also less prone to the polymer fouling.

AMBERLYST™ 36DRY is mainly used in the phenol alkylation processes.

Applications

· Phenol alkylation

Typical Properties

| Physical Properties | |
|-------------------------------|--------------------------------|
| Copolymer | Styrene-divinylbenzene |
| Matrix | Macroporous |
| Туре | Strong acid cation |
| Functional Group | Sulfonic acid |
| Physical Form | Black, opaque, spherical beads |
| Nitrogen BET | |
| Surface Area | 33 m ² /g |
| Total Pore Volume | 0.20 cc/g |
| Average Pore Diameter | 240 Å |
| Chemical Properties | |
| Ionic Form as Shipped | H ⁺ |
| Concentration of Acid Sites ‡ | ≥ 5.40 eq/kg |
| Catalyst Volatiles | ≤ 1.65% |
| Particle Size § | |
| Particle Diameter | 550 – 700 μm |
| < 300 µm | ≤ 2.0% |
| > 1180 µm | ≤ 3.0% |
| Swelling (in solvent) | |
| Phenol | 20% |
| Density | |
| Shipping Weight | 770 g/L |

[‡] Dry Weight Capacity ≥ 5.40 eq/kg

Suggested Operating Conditions

| Maximum Operating Temperature | 150°C (300°F) |
|-------------------------------------|--------------------------------|
| Bed Depth, min. | 600 mm (2.0 ft) |
| Pressure Drop, max. | 1 bar (15 psig) across the bed |
| Flowrates | |
| Linear Hourly Space Velocity (LHSV) | 0.5 – 5 h ⁻¹ |

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[§] For additional particle size information, please refer to the <u>Particle Size Distribution Cross Reference Chart</u> (Form No. 177-01775).

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Please be aware of the following:

 WARNING: Oxidizing agents such as nitric acid attack organic ion exchange resins under certain conditions. This could lead to anything from slight resin degradation to a violent exothermic reaction (explosion). Before using strong oxidizing agents, consult sources knowledgeable in handling such materials.

Have a question? Contact us at:

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