

Product Data Sheet

AMBERLITE™ FPA51 Ion Exchange Resin

Food-grade, Macroporous, Weak Base Anion Exchange Resin

Description

AMBERLITE™ FPA51 Ion Exchange Resin is a macroporous, weakly basic, anion exchange resin containing tertiary amine functionality on a crosslinked polystyrene matrix. It has been specifically designed for the purifying liquid food streams including starch-based sweeteners and can also be used in bioprocessing applications.

Nutrition Applications

AMBERLITE™ FPA51 Ion Exchange Resin has been sized to be used both in the fixed bed system commonly used in the corn sweetener industry as well as moving bed systems and polishing mixed bed. It is suitable for the deashing, deacidification, and decolorization of glucose, fructose and related starch-based sweeteners and derivatives as well as gelatin and other food process streams such as fruit juices.

Its high level of porosity gives AMBERLITE™ FPA51 an excellent combination of physical stability and high operating efficiency resulting in long process cycle times compared to products having a higher static volume capacity. This porous network also provides a more complete adsorption and desorption of large organic molecules resulting in excellent color removal compared to other weak base anion exchange resins.

Bioprocessing Applications

A number of different antibiotic classes have been isolated, chemically modified, and used extensively by physicians in treating infectious diseases. As most traditional antibiotics were derived from yeast or bacteria, their large-scale production is based on fermentation processes. AMBERLITE™ FPA51 Ion Exchange Resin can be used for the removal of organic color bodies in those downstream bioprocesses.

Applications

- Nutrition applications
 - Sweetener deashing
 - Sweetener deacidification
 - Sweetener decolorization
- Bioprocessing applications
 - Decolorization

Typical Properties

| Physical Properties | |
|--------------------------|--------------------------------|
| Copolymer | Styrene-divinylbenzene |
| Matrix | Macroporous |
| Туре | Weak base anion |
| Functional Group | Secondary amine (≥ 85%) |
| Physical Form | Beige, opaque, spherical beads |
| Chemical Properties | |
| Ionic Form as Shipped | Free base (FB) |
| Total Exchange Capacity | ≥ 1.3 eq/L |
| Water Retention Capacity | 56 – 62% |
| Particle Size § | |
| Particle Diameter | 490 – 690 μm |
| < 300 μm | ≤ 1.0% |
| > 1180 µm | ≤ 2.0% |
| Stability | |
| Swelling | FB → HCI : 25% |
| Density | |
| Shipping Weight | 660 g/L |
| | |

 $[\]S$ For additional particle size information, please refer to the <u>Particle Size Distribution Cross Reference Chart</u> (Form No. 177-01775).

Suggested Operating Conditions

| Maximum Operating Temperature | 100°C (212°F) |
|-------------------------------|--|
| Bed Depth, min. | 700 mm (2.3 ft) |
| Flowrates | |
| Service | 2 – 6 BV*/h (depending on syrup concentration) |
| Backwash | See Figure 1 |
| Regeneration | 1 – 2 BV/h |
| Slow Rinse | Regeneration flowrate for 2 BV |
| Fast Rinse (if applicable) | Service flowrate for 5 – 10 BV (with condensate or |
| | softened water) |
| Contact Time | |
| Regeneration | ≥ 30 – 45 minutes |
| Regenerant | NaOH |
| Concentration | 4% |
| Level | 60 kg/m³ (3.8 lb/ft³) |

^{* 1} BV (Bed Volume) = 1 m³ solution per m³ resin or 7.5 gal per ft³ resin

Hydraulic Characteristics

Estimated bed expansion of AMBERLITE™ FPA51 Ion Exchange Resin as a function of backwash flowrate and temperature is shown in Figure 1.

Estimated pressure drop for AMBERLITE™ FPA51 as a function of service flowrate and viscosity is shown in Figure 2. These pressure drop expectations are valid at the start of the service run with clean feed and a well-classified bed.

Figure 1: Backwash Expansion

Temperature = 5 - 60°C (41 - 140°F)

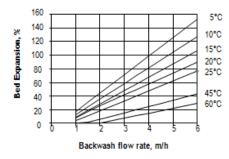
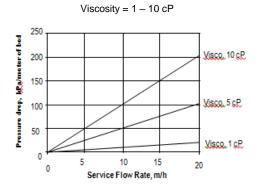


Figure 2: Pressure Drop



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Please be aware of the following:

WARNING: Oxidizing agents such as nitric acid attack organic ion exchange resins
under certain conditions. This could lead to anything from slight resin degradation to
a violent exothermic reaction (explosion). Before using strong oxidizing agents,
consult sources knowledgeable in handling such materials.

Have a question? Contact us at:

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