

Product Data Sheet

# AMBERSEP<sup>™</sup> 91419 and AMBERSEP<sup>™</sup> 91419 XL Chelating Resins

Gold-Selective Strong Base Anion Exchange Resin for the Recovery of Gold from Cyanidation Leach or Acid Chloride Leach Solutions

## Description

Today, most gold is mined by the cyanide leaching, or cyanidation, process. Typically, gold is associated with cyanide-soluble copper minerals that can lead to some operational challenges during the gold extraction process. The advantages of using selective anion exchange resins to sorb the gold from the copper-containing pregnant leach solution are broadly recognized, making ion exchange more cost efficient than other alternative routes like activated carbon.

AMBERSEP<sup>™</sup> 91419 and AMBERSEP<sup>™</sup> 91419 XL Chelating Resins are gold-selective resins with a t-butylamine functional group. The rapid kinetics of these chelating resins help to improve the recovery of gold, particularly in the presence of carbonaceous pregrobbing ores that otherwise preferentially absorb gold and gold-cyanide complexes. Both resins also feature a very strong mechanical stability, reducing the generation of fines during the processing steps and consequently enhancing the gold recovery efficiency by minimizing Au-loaded resin losses.

**AMBERSEP™ 91419 Chelating Resin**, with its standard particle size, is designed for systems with fixed or fluidized beds.

AMBERSEP<sup>™</sup> 91419 XL Chelating Resin, with its larger uniform particle size, is designed specifically for use in Resin-In-Pulp (RIP) processing, enabling an easy separation of the resin from the pulp. It is also a good choice for Resin-In-Leach (RIL) processes.

## Applications

- Gold recovery from cyanide leach
- Separation of gold from PGM streams

### **Typical Properties**

#### **Physical Properties**

| nyelean repetite |   |  |
|------------------|---|--|
| Copolymer        | Styrene-divinylbenzene                      |  |
| Matrix           | Macroporous                                 |  |
| Туре             | Chelant                                     |  |
| Functional Group | Quaternary amine (t-butylamine)             |  |
| Physical Form    | White to tan, hard, opaque, spherical beads |  |
|                  |   |  |

|                            | AMBERSEP™ 91419  | AMBERSEP™ 91419 XL |
|----------------------------|------------------|--------------------|
| Chemical Properties        |                  |                    |
| Total Exchange Capacity    | 0.23-0.33 eq/L   | 0.30-0.40 eq/L     |
| Dry Weight Capacity        | 0.8 – 1.2 meq/g  | 0.8 – 1.2 meq/g    |
| Water Retention Capacity   | 49-59%           | 45-55%             |
| Particle Size <sup>§</sup> |                  |                    |
| Particle Diameter          | 760 – 1200 μm    | 822 – 1445 μm      |
| Fine Beads                 | < 768 µm : ≤ 5%  | < 822 µm : ≤ 5%    |
| Coarse Beads               | > 1190 µm : ≤ 2% | > 1445 µm : ≤ 2%   |
| Density                    |                  |                    |
| Particle Density           | 1.08 g/mL        | 1.08 g/mL          |
| Shipping Weight            | 670 g/L          | 670 g/L            |

§ For additional particle size information, please refer to the <u>Particle Size Distribution Cross Reference Chart</u> (Form No. 177-01775).

# Suggested Operating Conditions

| Maximum Operating Temperature |                     |
|-------------------------------|---------------------|
| OH⁻form                       | 60°C (140°F)        |
| Cl <sup>-</sup> form          | 100°C (212°F)       |
| pH Range                      | 0 - 14              |
| Bed Depth, min.               | 800 mm (2.6 ft)     |
| Total Rinse Requirement       | 2-4 BV*             |
| Regenerant                    | Thiourea, acidified |

\* 1 BV (Bed Volume) = 1  $\text{m}^3$  solution per  $\text{m}^3$  resin or 7.5 gal per ft<sup>3</sup> resin

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Please be aware of the following:

• **WARNING:** Oxidizing agents such as nitric acid attack organic ion exchange resins under certain conditions. This could lead to anything from slight resin degradation to a violent exothermic reaction (explosion). Before using strong oxidizing agents, consult sources knowledgeable in handling such materials.

Have a question? Contact us at:

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